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Novel Synthesis of Castanospermine and 1-Epicastanospermine

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ABSTRACT

Polyhydroxylated indolizidines have potential for treatment of HIV, hepatitis C and HSV infection, multiple sclerosis, angiogenesis, cancer, and diabetes. A new synthetic approach to the title compounds from a 5-C-methoxypyranosyl azide has been developed. The route incorporates the aldol reaction and a novel catalytic reductive amination cascade to generate the indolizidine ring.

Castanospermine **1a** is a bioactive naturally occurring polyhydroxylated indolizidine, first isolated from seeds of *Castanospermum australe*.¹ It and closely related congeners have shown considerable potential as antiviral agents for the treatment of HIV,² hepatitis C,³ and HSV-1⁴ infections. They also have potential for inhibition of progression of multiple sclerosis,⁵ angiogenesis, cancer,⁶ and diabetes.⁷ The development of new synthetic routes to castanospermine enable the production of novel analogues for biological studies, and a

number of syntheses of **1** have been developed.⁸ Herein we present a novel synthesis of castanospermine **1a** and 1-epicastanospermine **1b** from methyl α -D-glucopyranoside.

We have recently reported a general synthesis of iminosugars from 5-*C*-methoxypyranosyl azides (e.g., production of 3 from 2). We were interested to investigate extending the catalytic reductive amination cascade reaction used for

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preparation of piperidine 3 to generate the indolizidine ring system (Scheme 1). It was envisaged that the piperidine 5

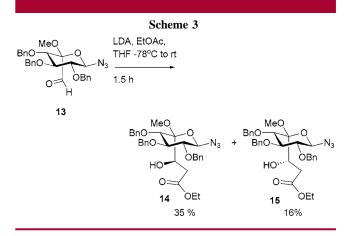
generated from 5-*C*-methoxypyranosyl azide **6** would undergo an additional cyclization and give the indolizidinone **4**, which would be subsequently converted to **1a**.

The synthesis commenced from the 6-deoxyhex-5-enopyranoside **7**, prepared from methyl α -D-glucopyranoside in five steps (69%) as previously described. 9b The acetate groups

were removed, and benzyl protecting groups were introduced to give **8**. Oxidation using methyl(trifluoromethyl)dioxirane generated in situ¹⁰ gave a 1.7:1 mixture of epoxides **9**.¹¹ Treatment of the epoxides with methanol in the presence of camphorsulfonic acid gave a mixture of 5-*C*-methoxy-D-glucopyranosyl azide **10** and 5-*C*-methoxy-L-idopyranosyl azide **11**. This mixture was oxidized to aldehydes **12** and **13** using the method of Ley and Griffith.¹² The aldehydes were separated using chromatography,¹³ and the configurations assigned to **12** and **13** were determined using 1D NOE spectroscopy. An NOE enhancement was observed for H-3 on irradiation of the aldehyde proton for **13**, with no enhancement observed for H-4. An NOE enhancement was observed for H-4 of **12** on irradiation of its aldehyde proton with no enhancement observed for H-3 (Figure 1). The aldol

Figure 1. Assignment of the configuration of 12 and 13

reaction of 13 was next investigated. Reaction of ethyl acetate at -78 °C with LDA generated the desired enolate, which



on reaction with 13 gave a mixture of diastereoisomeric alcohols 14 and 15; these were separated using silica gel

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⁽¹³⁾ **Oxidation to 12 and 13.** To a mixture of **10** and **11** (1.6 g, 3.2 mmol) in dry CH_2Cl_2 (35 mL) were added activated 4 Å molecular sieves (0.1 g) and *N*-methylmorpholine *N*-oxide (NMO, 0.56 g, 4.8 mmol). After 30 min of stirring at room temperature under nitrogen, tetra-*n*-propylammonium perruthenate (TPAP, 0.11 g, 0.31 mmol) was added, and the mixture was stirred overnight and then filtered through Celite. Filtration through a short column of silica gel, eluting with CH₂Cl₂ gave **13** (0.74 g, 46%). Further elution with EtOAc gave **12** (0.33 g, 21%) as a clear oil.

chromatography with **14** being eluted first. The one-pot reductive indolizidine ring generating cascade reaction was then investigated and, gratifyingly, succeeded in a highly stereoselective manner for both substrates **14** and **15**, giving lactams **16** (70%) and **17**¹⁴ (62%), respectively. ¹⁵ Reduction of **16** and **17** was achieved by protection of the hydroxyl groups as trimethylsilyl ethers and subsequent reaction using LiAlH₄ in THF, giving 1-epicastanospermine **1b** and (+)-castanospermine **1a**, respectively. ¹⁶ Indolizidine **1a** was converted to its per-*O*-acetate, which was subjected to chromatographic purification, and subsequent deacetylation

gave a sample of **1a** of higher purity. This sample was found to have ¹H and ¹³C NMR spectroscopic data¹⁷ identical to those previously reported¹⁸ and to those of an authentic sample. Similarly **1b** was found to have spectroscopic data identical to those reported in the literature.¹⁹ A possible sequence of intermediates explaining the formation of the lactam **17** is shown in Scheme 5. This involves reduction of

azide group to give a pyranosylamine, which ring opens to an acyclic imine. Further reductive cyclization gives a cyclic imine, which is reduced in situ to give 5, and subsequent cyclization gives lactam 17.

In summary, a new strategy for the synthesis of polyhydroxylated indolizidines and related lactams has been demonstrated and should be amenable to the synthesis of further analogues. A wider investigation of the aldol reaction from the view of generating analogues for biological evaluation as well as improving stereoselectivity would seem appropriate next studies. In addition aldehydes such as 13 could have more wide application as synthetic intermediates to novel iminosugars and indolizidines.

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Supporting Information Available: Full experimental details and ¹H and ¹³C spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

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⁽¹⁵⁾ **Preparation of 17.** Azide **15** (110 mg, 0.186 mmol), Pd(OH)₂ (30 mg), and formic acid (0.82 mL) were added to MeOH (60 mL), and the mixture was stirred in a Parr reactor in an atmosphere of hydrogen (500 psi) for 48 h. The suspension was then filtered through Celite, washing first with MeOH and then water; the combined filtrate was evaporated to dryness in vacuo; and the residue was purified using silica gel chromatography (4:1 EtOAc/MeOH) to give **17** (23 mg, 62%).

⁽¹⁶⁾ The removal of the TMS protecting groups is facile and occurs during the workup after reduction.

⁽¹⁷⁾ $^{1}\mathbf{H}$ NMR data for 1a: δ (300 MHz, D₂O) 4.41 (ddd, 1H, J 6.8, J 4.6, J 1.5), 3.61 (m, 1H, H-6), 3.60 (t, 1H, J 9.6), 3.32 (t, 1H, J 9.1), 3.18 (dd, 1H, J 10.8, J 5.1), 3.08 (dt, 1H, J 9.1, J 2.1), 2.33 (dddd, 1H, J 14.8, J 8.1, J 7.3, J 2.2), 2.21 (q, 1H, J 9.2), 2.06 (t, 1H, J 10.7), 2.02 (dd, 1H, J 9.8, J 4.4), 1.17 (dddd, 1H, J 14.2, J 8.7, J 8.7, J 1.8).

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